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# Molecular Crystals and Liquid Crystals Incorporating Nonlinear Optics

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# Properties of the Liquid Crystals Formed by Certain Azomethines Derived from 4-Cycloalkylanilines and from 4-Cycloalkylbenzaldehydes

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Mol. Cryst. Liq. Cryst., 1988, Vol. 159, pp. 37-52 Photocopying permitted by license only © 1988 Gordon and Breach Science Publishers S.A. Printed in the United States of America

# Properties of the Liquid Crystals Formed by Certain Azomethines Derived from 4-Cycloalkylanilines and from 4-Cycloalkylbenzaldehydes†

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The liquid crystal behaviour of four homologous series of azomethines related to the nO.m series but containing a cycloalkyl group, is reported and discussed. Many of these compounds show extensive smectic polymorphism, one member of the nO.c6 series giving rise to five smectic polymorphic modifications for which phase type assignments have been made by thermal optical microscopy.

Keywords: smectic polymorphism, azomethines, cycloalkyl derivatives

<sup>†</sup>Presented at the 11th International Liquid Crystal Conference, University of California, Berkeley, California, 30th June-4th July, 1986.

## INTRODUCTION

The liquid crystal behaviour is reported of members of the series of N-(4-n-alkoxybenzylidene)-4'-cycloalkylanilines (1) derived from 4-cyclopentyl- and -cyclohexyl-aniline, and the 'reversed' series, the 4-n-alkoxy-N-(4'-cycloalkylbenzylidene)anilines (2), derived from 4-cyclopentyl- and -cyclohexylbenzaldehyde. These azomethines are related to the nO.m series¹ [of which the compound (3), MBBA or 1O.4 is a simple member].

Ar—N=CH—
$$\bigcirc$$
OC<sub>n</sub>H<sub>2n+1</sub> Ar—CH=N— $\bigcirc$ OC<sub>m</sub>H<sub>2m+</sub>
(1)
(2)
CH<sub>3</sub>O— $\bigcirc$ CH=N— $\bigcirc$ C<sub>4</sub>H<sub>9</sub>
(3) MBBA, 1O.4

where Ar =  $\bigcirc$  and Ar =  $\bigcirc$ 

By analogy with the nO.m nomenclature, the **cyclo**alkyl substituted azomethines (1) are conveniently described as members of the nO.cm series and those of 'reversed' type (2) belong to cn.Om series.

Initially, our interest was only with series (1): Ar = 4-cyclohexylphenyl, the intention being to compare the known effect<sup>2</sup> of Ar = 4-biphenylyl with its 'half-reduced' Ar analogue. However, study of a range of homologues (n = 1-10, 12, 14, 16) of this nO.c6 series revealed very extensive smectic polymorphism reminiscent of the corresponding smectogenic nO.6 series. One member (n = 12) of the nO.c6 series gives rise to a nematic phase and five smectic polymorphic modifications for which phase type assignments have been made based on the textures observed by thermal optical microscopy. Related homologues, (1): Ar = cyclopentylphenyl (nO.c5) and those of the 'reversed' series (2): Ar = 4-cyclohexylphenyl (c6.Om) and Ar = 4-cyclopentylphenyl (c5.Om) show similar, though less pronounced smectic polymorphism.

### **RESULTS AND DISCUSSION**

Transition temperatures for eleven or twelve [(n, m = 1 (or 2) - 10, 12, 14] or thirteen (also n = 16) members of each of the four series,

TABLE I

Transition temperatures for nO.c6 series; (1): Ar = 4-cyclohexylphenyl

		4				6 16		
<b>c</b>	C-I(N, S <sub>A</sub> )	$N(S_A)-I$	$S_A(S_C, S_G)$ –N	$S_C(S_B)-S_A$	$S_G(S_B)-S_C$	$S_{F}$ – $S_{B}$	$S_G$ - $S_F$	$N(S_G, S_F)$ –C
	C-I	I-N		į				N-C
_	107	(92.5)						74.5
7	139	(118.5)						114
e	107	(95)						94
4	118	(105.5)						85.5
	Z-J							
5	88.5	%						75
			$S_{G}$ N					S <sub>G</sub> -C
9	84.5	102	(73)					72.5
7	92	97.5	(71.5)					99
			S <sub>C</sub> -N		$S_{G}$			
<b>%</b>	79.5	100	(75)		(71)			38
			S <sub>A</sub> -N	$S_{C}-S_{A}$				
6	95.5	97.5	(83)	(78.5)	(69)			39
5	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	٤	2 00	(3 02)	(3 03)			6
01	63.3	86	67.70	(6.67)	(00.3) S <sub>B</sub> -S <sub>C</sub>			<b>.</b>
12	87	97.5	94.5	(92)	(70.5)	(70)	(67.5)	39
		$S_{A}$ -I		$S_B - S_A$				
14	88	95.5		(72.5)		(65.5)	(63.5)	52.5
16	87.5	94		(75)		(71)		26.5
				1997				

( ) monotropic transition

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TABLE II

		Transition temp	peratures for nO.c5	Transition temperatures for nO.c5 series; (1): Ar = 4-cyclopentylphenyl	cyclopentylpheny	71	
u	$C-I(N, S_G, S_A)$	N(S <sub>A</sub> )-I	S <sub>A</sub> (S <sub>G</sub> )–N	$S_G-S_A(S_B)$	S <sub>B</sub> -S <sub>A</sub>	S <sub>F</sub> -S <sub>B</sub>	$N(I, S_G, S_F)$ –C
	C-I	I-N					N-C
-	% \  % \  % \	(84)					82
7	104.5 C I	110					76
т	C-7						N-01
4	95 S-7	86	Z   V				92 S-C
5	69.5	88	79				\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
9	78	94.5	84				<25
7	57.5	90.5	79				<25 <u>-</u>
<b>∞</b>	09	94	% <u>%</u>	y I			<25
6	63	91.5	82.5	75.5			31
10	19	93	<u>%</u>	S <sub>G</sub> -S <sub>B</sub>			33
12	73.5	92 S <sub>2</sub> –I	91.5	73.5	76.5		43
14	92	91.5		(68.5)	77.5		53.5
16	84	06			(79.5)	(70)	68
ouom ( )	( ) monotropic transition			-			i

TABLE III
Transition temperatures for c5.Om series; (2): Ar = cyclopentylphenyl

m	$C-I(N, S_G)$	N-I	$S_G(S_A)-N$	$N(S_G, S_A)$ -C
	C-I			N-C
1	$\frac{\text{C-I}}{93}$	(87.5)		$\frac{N-C}{68}$
	C-N	,	$S_G-N$	$S_G-C$
2	107.5	113	$\frac{S_{G}-N}{(94.5)}$	80
	<u>C-I</u> 94 <u>C-S<sub>G</sub></u>		. ,	
3	94	(91)	(87)	78
	C-S <sub>G</sub>	` '	•	
4	80	102.5	89.5	75
	C-N			
5	83.5	93	84	58.5
6	89.5	98.5	(87)	77.5
7 8 9	85	93.5	(81.5)	67.5
8	88	95.5	(81.5)	76.5
9	86	92	(74.5)	74
10	86.5	93	(73.5)	73
			` '	N-C
12	89.5	91		76
	C-I		$S_A-I$	$S_A-C$
14	91.5	88.5	(79)	78.5

( ) monotropic transition

(1) and (2): Ar = 4-cyclopentylphenyl and Ar = 4-cyclohexylphenyl, are listed in Tables I–IV and shown plotted against the number of carbon atoms (n,m) in the alkyl chain of the alkoxy group in Figures 1–4.

The results show clear similarities between the liquid crystal behaviour of the nO.5 and nO.6 series and their cycloalkyl analogues, the nO.c5 and nO.c6 series. The presence of the cycloalkyl group does, however, shift the onset of this behaviour to longer alkoxy chain lengths, and many of the smectic phases are monotropic as m.p.s. are higher in the nO.c5 and nO.c6 series. Comparison of the average values of  $T_{\rm N-I}$  for n = 4-7 shows that the nematic thermal stabilities of the nO.6 and nO.5 series are lower than those of their cycloalkyl analogues, nO.c6 and nO.c5, by 22.9°C and 10.2°C, respectively.

There are no data on n.Om series for comparison with the c5.Om and c6.Om series. However, for these cycloalkyl analogues, reversal of the azomethine linkage results in a similar pattern of behaviour except that smectic polymorphism is less extensive with its onset at higher homologues than in the nO.cm series. The compounds, in general, do not allow the same degree of supercooling as those of the nO.cm series.

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TABLE IV

	I(N, S <sub>c</sub> , S <sub>G</sub> )-C	N-C 86.5 104 104 104	101	80.5	74 76 75 70 70 70 70 70 70 70 70 70 70 70 70 70	65 % C	74.5
4-cyclohexylphenyl	$S_A(S_G)$ – $S_C$					$\frac{S_G - S_C}{(67)}$	(75.5)
Transition temperatures for c6.0m series; (2): Ar = 4-cyclohexylphenyl	$S_A(S_C, S_G)$ –N			N-S	(75) (71) (72) (72) (72)	$S_{C-N}^{(73.3)}$	(84)
Transition temperatures	I-N	120.5	(110.5) (101.5)	106.5	101	101 98.5	95.5
	C-I(N)	C-N 116.5 C-1 121	117.5 104.5 O_N	92.5	98.5 98.5 91	91.3 98.5 7-1	97.5

9

( ) monotropic transition

12

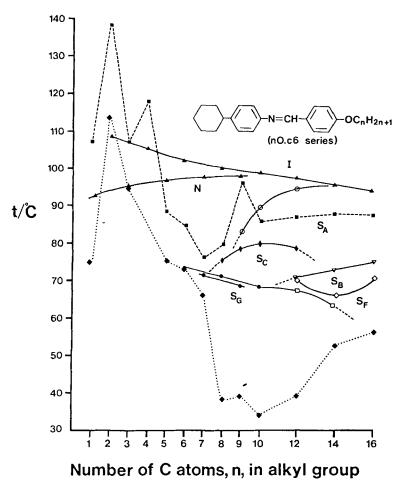


FIGURE 1 Plot of transition temperatures against the number of carbon atoms (n) in the alkyl chain of the N-(4-n-alkoxybenzylidene)-4'-cyclohexylanilines (nO.c6 series). ■---■ m.p.s., ♦ --- ♦ --- ♦ , recrystallisation temperatures.

# **Optical textures**

The various smectic polymorphic modifications were identified by thermal optical microscopy. Optical textures shown by the liquid crystal phases of the compound n=12 of the nO.c6 series, N-(4-n-dodecylbenzylidene)-4'-cyclohexylaniline, are shown in the photomicrographs, Plates 1–6. The section of the sample is the same in each instance and contains a region in which orthogonal smectic phases

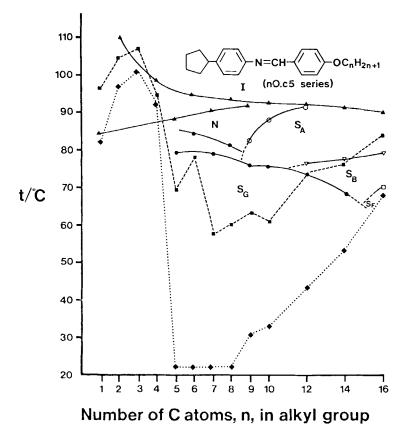


FIGURE 2 Plot of transition temperatures against the number of carbon atoms (n) in the alkyl chain of the N-(4-n-alkoxybenzylidene)-4'-cyclopentylanilines (nO.c5 series).

appear homeotropic. The smectic phase types assigned as  $S_A$ ,  $S_C$ ,  $S_B$  and  $S_G$  show textural characteristics which are typical of these phases as described in the texts by Gray and Goodby<sup>3</sup> and by Demus and Richter.<sup>4</sup> Similarly, the smectic phase that occurs intermediate between the  $S_B$  and  $S_G$  phases shows optical textures consistent with it being an  $S_F$  phase. At the transition to this phase on cooling the  $S_B$  phase, the previously smooth  $S_B$  fans become broken at their edges and then take on a mottled appearance with irregular birefringent colouring. Also the homeotropic region of the previous  $S_B$  phase is invaded, at first by a string-like interlocking network which lacks colour. This grows until the whole region is occupied by very small mosaic or *schlieren*-mosaic areas (rather than the *schlieren* texture

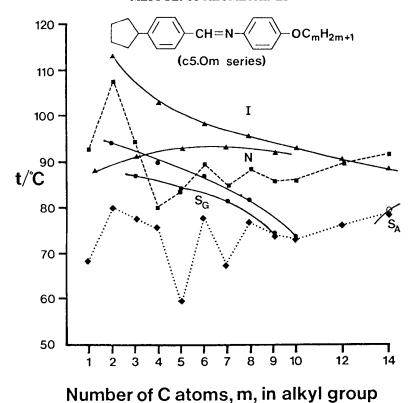


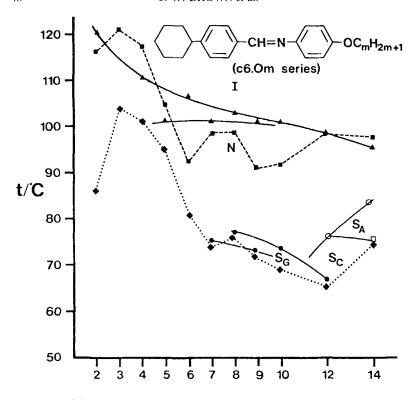
FIGURE 3 Plot of transition temperatures against the number of carbon atoms (m) in the alkyl chain of the 4-n-alkoxy-N-(4'-cyclopentylbenzylidene)anilines (c5.Om series).  $\blacksquare$ --- $\blacksquare$  m.p.s.,  $\blacklozenge$ --- $\blacklozenge$ --- $\blacklozenge$ -, recrystallisation temperatures.

that might have been expected for an  $S_1$  phase). These textures are exactly analogous to those shown by the  $S_F$  phase of the compound  $50.6.^1$  We consider, therefore, that the compounds 50.6 and 120.c6 show the same sequence of phases on cooling the isotropic liquid, namely  $I-N-S_A-S_C-S_B-S_F-S_G$ . Thus, as with the compound 50.6 the  $S_F$  phase of the compound 120.c6 occurs between the *more ordered*  $S_B$  and  $S_G$  phases.

The assignments of phase type by thermal optical microscopy will be confirmed by miscibility studies and by X-ray diffraction.

## Characteristics of individual series

Some similar trends are apparent in all four series studied. Thus, as n(m) increases: (i) the onset of smectic properties is marked by the



# Number of C atoms, m, in alkyl group

FIGURE 4 Plot of transition temperatures against the number of carbon atoms (m) in the alkyl chain of the 4-n-alkoxy-N-(4'-cyclohexylbenzylidene)anilines (c6.Om series). ■---■ m.p.s., ♦ --- ♦ --- ♦ , recrystallisation temperatures.

occurrence of an  $S_G$  phase, (ii)  $S_G$ -N transition temperatures show odd-even (n,m) alternation, and (iii) N-I transition temperatures for even-n(m) members fall and those of odd-n(m) members initially rise.

nO.c6 series; (1): Ar = 4-cyclohexylphenyl (Figure 1, Table 1). The early members are purely nematic. Smectic properties appear with an  $S_G$  phase at n = 6. At n = 8, on raising the temperature, the  $S_G$  phase gives rise to another tilted smectic, an  $S_C$  phase, before giving way to the nematic phase. For n = 9 and 10, there is similar behaviour except that an  $S_A$  phase is interposed between the  $S_C$  and the nematic phases. The  $S_G$ - $S_C$  transition temperature curve rises to n = 10 then falls to n = 12, whereas the  $S_A$ -N transition temperatures continue to rise so that, for n = 14 and 16,  $S_A$ -N transitions replace the N-I transitions of earlier members.

PLATES 1-6. Photomicrographs of the optical textures shown by the liquid crystal phases of N-(4-n-dodecylbenzylidene)-4'-cyclohexylaniline (12O.c6) on cooling from the isotropic liquid. The section of the sample is the same in each instance and contains a region in which orthogonal smectic phases appear homeotropic.



PLATE 1 Nematic phase. Schlieren texture on right hand side.



PLATE 2 Smectic A phase. Focal conic fan texture with a homeotropic region in the schlieren area of the previous nematic phase.



PLATE 3 Smectic C phase. Broken focal conic fan texture with a schlieren region in the homeotropic area of the previous  $S_A$  phase.

For n=12, 14 and 16, the smectic polymorphism becomes rather complex. For n=12, on raising the temperature, the  $S_G$  phase gives successively an  $S_F$  phase then an  $S_B$  phase (which persist for 2.5° and 0.5°C, respectively) before the  $S_C$  phase is formed, so that the sequence of phases is C,  $S_G$ ,  $S_F$ ,  $S_B$ ,  $S_C$ ,  $S_A$ ,  $S_B$ ,  $S_C$ ,  $S_A$ ,  $S_B$ ,  $S_C$ 

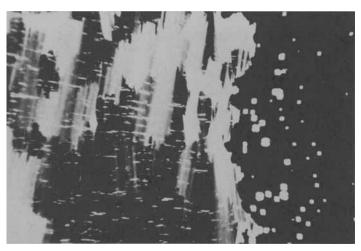


PLATE 4 Smectic (crystal) B phase. The schlieren area of the previous  $S_C$  phase has become homeotropic. The transition bars at the  $S_C$ - $S_B$  transition have left a number of 'fissures' in the fans. These slowly disappear on standing.

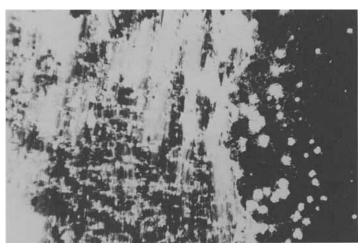


PLATE 5 Smectic F phase. The fans of the previous  $S_B$  phase have a broken, mottled appearance and a network of very small mosaic areas is developing in the homeotropic region of the previous  $S_B$  phase. This gradually fills the entire region.

tallisation obscures the formation of the  $S_G$  phase on cooling the  $S_F$  phase.

nO.c5 Series; (1): Ar = 4-cyclopentylphenyl (Figure 2, Table II). This series shows very similar trends to the nO.c6 series except that no  $S_C$  phases are evident. The early members are purely nematic.  $S_G$  phases appear at n = 5, and, as in the nO.c6 series,  $S_A$  phases

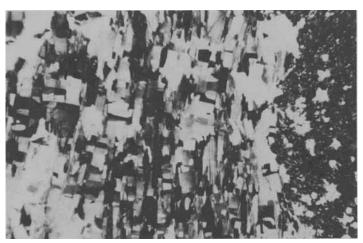


PLATE 6 Smectic G phase. Broken focal conic fan and mosaic textures. The fans have a characteristic patchwork or chequerboard appearance.

The cyclohexyl analogues were prepared similarly

$$CH=N- OC_m H_{2m+1} OC_n H_{2m+1}$$

$$C6.Om series OC_n H_{2m+1}$$

FIGURE 5 Synthetic scheme used for the preparation of the N-(4-n-alkoxybenzylidene)-4'-cycloalkylanilines (1) (nO.cm series) and the 4-n-alkoxy-N-(4'-cycloalkylbenzylidene)anilines (2) (cn.Om series).

occur at n = 9, the  $S_A$ -N transition temperatures lying on a rising curve that becomes coincident with the falling N-I curve at n = 14. For n = 12, 14 and 16, the  $S_A$  phase gives rise to an  $S_B$  phase on cooling so that for n = 12 and n = 14,  $S_G$ - $S_B$  transitions occur, although for n = 16, the  $S_G$  phase is absent and there is an  $S_F$ - $S_B$  transition instead.

c5. Om Series; (2): Ar = 4-cyclopentylphenyl (Figure 3, Table III). These compounds show nematic phases throughout (m = 1 to m = 14).  $S_G$  phases commence at n = 2 and the  $S_G$ -N transition temperature curves fall up to m = 10. For m = 9 and m = 10, recrystallisation occurs immediately after the N- $S_G$  transition, and for the compound m = 12, recrystallisation extinguishes smectic properties. These reappear at m = 14 when an  $S_A$ -N transition occurs, presumably at the start of a curve rising to meet the N-I curve although higher members than n = 14 were not studied.

c6.Om Series; (2): Ar = 4-cyclohexylphenyl (Figure 4, Table IV). This series resembles the c5.Om series except that the early members have rather high recrystallisation temperatures which obscure the  $S_G$  phases so that  $S_G$ -N transitions do not appear until n = 7. In contrast with the c5.Om series, the compound n = 12 has a lower temperature of recrystallisation and shows both  $S_G$  and  $S_C$  phases. For n = 14, an  $S_A$  phase lies above the  $S_C$  phase, but the  $S_G$  phase does not occur.

Work on other cycloalkyl analogues is in progress.

#### EXPERIMENTAL

### Materials

The azomethines were prepared by standard methods illustrated by the routes to the 4-cyclopentylphenyl derivatives, the 4-n-alkoxy-n-(4'-cyclopentylbenzylidene)anilines (c5.Om series) and N-(4-n-alkoxybenzylidene)-4'-cyclopentylanilines (nO.c5 series), shown in the synthetic scheme, Figure 5. Full experimental details of the synthetic work (which may be obtained on request to the authors, DJB or RCW) will appear in a future publication.

#### Physical measurements

Measurements of transition temperatures and microscope observations of textures of mesophases were made using a Vickers M75 polarising microscope in conjunction with a Mettler FP 52 hot-stage and FP 5 control unit.

Structural confirmation of the identity of intermediates (where necessary) and final products was obtained by standard techniques. I.r. spectra were recorded for KBr discs with a Perkin-Elmer 157 grating spectrophotometer, and <sup>1</sup>H n.m.r. spectra were measured for solutions in CDCl<sub>3</sub> with tetramethylsilane as internal standard with a JEOL FX60Q Fourier transform spectrometer. Mass spectra were determined with an A.E.I. MS 902S spectrometer equipped with a Mass Spectrometry Services Analog Series 200 console and an INCOS 2300 data system.

# **Acknowledgements**

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